EXPERIMENT Nº 01

EXPERIMENT USING GLASS CONTAINERS (I)

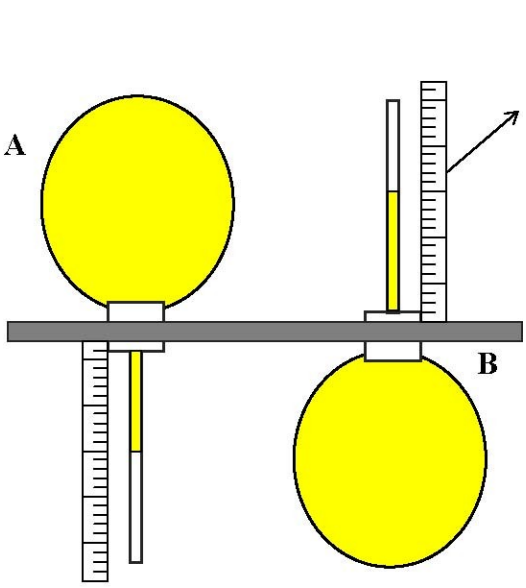

Illustration 01 - Reversed containers

OBJECTIVES:

In case there is an attraction force exerted by the moon over the earth, when the level of (B) reaches its maximum, the level of (A) should remain the same, or it could decrease the volume of the solution contains within the graded burette, the reason for that is that the liquid should become lighter due to the attraction exerted by the moon thereby allowing the cohesion force which exists between the molecules to regroup, coming closer together, so there will be a decrease in the volume of (A) which will also be caused by atmospheric pressure.

On the other hand, if there is a `spatial Flow`, the levels of a (A) and (B) should increase and decrease equally, that is because both volumes are the same and they are in the same height from the ground, therefore, and the amount of `Space` absorbed by (A) is equal to the amount of space absorbed by (B).

PROCEDURES:

Both containers made of stained glass containing the same solution with the same density. (30g of salt per litter), two graded 2ml burettes joined in the bottleneck via a Teflon cork. (If you are using non graded burettes, the use of rulers is necessary).

A and **B** = 1000ml of ***H2O*** + ***NaCl***

CONCLUSIONS:

If there is an attraction force exerted by the moon upon the earth capable of "raising" millions of tons of sea water forming the high tides, it should cause (proportionally) a variation in the weight of the liquid masses used in this experiment, therefore resulting in a predictable behavior on both levels, since all the relevant factors were known beforehand, factors such as electrostatic, thermodynamic and environmental (CNTP).

What transpired during experiment number 01 was the front then would be expected were there an "attraction force". What was verified instead were the `flows of spatial energy` which traverse the solutions and is partially absorbed. It generates work, raises `temperature` and the `volume` of the solutions also increases.

PS:
The graphic refers to Volume VII
Experimental database is in Volume VII

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